

NUCLEATED THERMOPLASTIC ELASTOMER COMPOSITIONS AND RELATED METHODS

5 Claim of Priority

 This application claims priority from U.S. Provisional Patent Application Serial Number 60/491,147 bearing Attorney Docket Number 12003009 and filed on July 30, 2003.

10 Field of the Invention

 This invention relates to use of nucleating agents in propylene-based thermoplastic elastomers.

Background of the Invention

15 In the past several decades, the use of polymers has transformed the world. Polymer science has rapidly evolved to make thousands of different thermoplastic and thermosetting products within the four corners of polymer physics: thermoplastic plastics, thermoplastic elastomers, thermoset plastics, and thermoset elastomers. Of these four, a thermoplastic elastomer (also
20 "TPE") exhibits both the valuable performance properties of an elastomer and the valuable processing properties of a thermoplastic.

 Thermoplastic elastomers are generally identified and categorized, as reported in Thermoplastic Elastomers, edited by Holden et al. (1996), based on the type of elastomer. A "thermoplastic elastomer" (TPE) is generally a
25 polymer or blend of polymers that can be processed and recycled in the same way as a conventional thermoplastic material, yet having properties and performance similar to that of an elastomer or rubber at the service temperature at which it is used. Notably, blends (or alloys) of plastic and elastomeric rubber have become increasingly important in the production of thermoplastic
30 elastomers, particularly for the replacement of thermoset rubber or flexible polyvinyl chloride (PVC) in various applications.

A "thermoplastic vulcanizate" (TPV) is a type of thermoplastic elastomer, where the elastomer phase is partially or completely crosslinked, vulcanized or cured, such that the TPV can be processed and recycled in the same way as a conventional thermoplastic material, yet retaining properties and performance similar to that of a vulcanized elastomer or rubber at the service temperature at which it is used. TPVs are becoming increasingly important in the production of high performance thermoplastic elastomers, particularly for the replacement of thermoset rubber in various applications.

No large scale production of any polymer can rest on current processing conditions. Reduction of cost, improvement of productivity, delivery of better performing, lower cost products all drive the polymer science industry. The situation is no different for TPEs.

Summary of the Invention

The art needs a means to enhance initiation and propagation of crystal growth in thermoplastic elastomers (TPEs), as a means to reduce production time and effort and increase productivity. The art also needs a means to improve the transparency of a TPE, while at the same time enabling efficient processing of compositions comprising TPEs.

The present invention solves the problem in the art by introducing into TPEs the facility of a nucleating agent to increase the number of nucleation sites in the thermoplastic phases of TPEs from which crystal growth propagates. By increasing the number of crystals formed within the thermoplastic phase of a TPE, the crystal structure of the TPE composition becomes solid more quickly upon cooling.

During processing of TPE articles, it is advantageous economically to get the TPE to solidify as quickly as possible without warping the article being formed. For example, when injection molding parts from the TPE, once a crystal structure is formed within the thermoplastic phase of a TPE, the article can be removed from the mold. This enables more parts to be produced per

hour – reducing the cycle time, and hence increasing the production rate and reducing the processing cost.

Not only has the introduction of nucleating agents into TPEs been discovered to increase productivity of TPE compositions and articles therefrom, but it also promotes formation of transparent TPE compositions, especially when the TPE is used to form relatively thin articles. In certain embodiments of the invention, as more nucleation sites from which crystal growth propagates are introduced, the resulting crystal size in the processed article tends to be smaller than without the introduction of those nucleation sites. This phenomenon facilitates transparency in the article depending on how it is otherwise processed (e.g., if the article is processed to a relatively low thickness).

The TPE composition of the invention is a combination of traditional TPEs with one or more nucleating agents. Unexpectedly, while nucleating agents have been known for use with non-self-nucleating polymeric resins, such as polypropylene, no one has seen fit to include such nucleating agents in the thermoplastic portion of a TPE as afforded by the present invention.

Unexpectedly, one can reduce production cycle times by as much as at least 25 percent (%) because the rate of initiation and propagation of crystal growth in the thermoplastic phase of a melt-processed thermoplastic elastomer compound of the present invention is increased by a like percentage. The producer of the final part or article from this melt-processed TPE composition can, therefore, increase the rate of cooling and completion of subsequent processing steps without complication by the effects of crystal growth after cooling of the material below the melting or softening temperature of the thermoplastic phase.

While these efficiencies are beneficial in processing TPE compositions for a variety of applications, the efficiencies are particularly noteworthy when forming the TPE compositions into final articles using processing techniques that heat the TPE composition to a temperature higher than the melting or

softening temperature of the thermoplastic phase. For example, injection molding and thermoforming are two such processes. A production rate limiting step in these types of processes is often the cooling step, whereby the processed material is transformed from the liquid or molten state to the solid state.

- 5 Because of the reduced time it now takes to crystallize and hence solidify the thermoplastic phase of TPE compositions according to the present invention, the material is able to cool more quickly. Advantageously, this increases the overall production efficiency.

- 10 Further, TPE compositions of the invention are prone less to the well-known phenomenon of super cooling. By minimizing super cooling within the TPE compositions, articles formed by the TPE compositions are able to advance through, for example, the quality control step of production more rapidly as the article's crystal structure and resulting properties stabilize much sooner than articles formed from conventional TPEs (i.e., without a nucleating agent). In
15 preferred embodiments of the invention, quality control is able to be reliably performed within no more than 2 hours, and even within minutes of an article's formation. In contrast, articles processed using conventional TPEs can take up to about 24 or even 72 hours to stabilize to the point where quality control analysis thereof can be reliably performed and the article can be used for its
20 intended application.

- The improvements in the art of TPE melt-processing apply to both the manufacturer of the TPE composition and to the customer of that TPE composition. More importantly, that customer who re-melt-processes the TPE composition can benefit directly from manufacturing efficiencies introduced by
25 way of the compositions of the present invention. The benefits seen by the end user are possible without the need for that user to introduce additional processing equipment or steps into what may already be a timely and complicated process.

- The introduction of the nucleating agent into the TPE compositions of
30 the present invention is not complicated and can utilize one of several methods:

(1) addition during formation of the TPE itself, (2) addition as a post-processing step after the TPE itself is formed but before the TPE is transferred to another melt-processor (e.g., the end user who will transform the TPE composition into a desired article), or (3) addition during final processing of the TPE into a part or article. In that respect, the second method choice is similar to what the last melt-processor could do itself as in the third method, but for the fact that if the nucleating agent is introduced at that stage, it typically does not get distributed evenly without a melt mixing or compounding step and further, the last melt-processor typically has a desire to streamline its own manufacturing steps.

Embodiments of the Invention

Thermoplastic elastomer (TPE) compositions of the invention comprise at least one TPE and at least one nucleating agent. The compositions comprise both of these components, before and after melt-processing, as the nucleating agent typically does not chemically react with the TPE nor degrade during processing.

Nucleating agents of the invention are useful for preparing unique TPE compositions comprising at least one elastomer phase and at least one thermoplastic phase. In further embodiments of the invention, the TPE compositions comprise at least two chemically distinct thermoplastic phases. In still further embodiments, the TPE compositions comprise at least two chemically distinct elastomer phases. In yet further embodiments, the TPE compositions also comprise optional processing oils, other additives, or combinations thereof, as well known to those skilled in the art.

Thermoplastic Elastomer (TPE)

A wide variety of TPEs can be used beneficially with nucleating agents of the invention. Within a TPE, there is at least one phase of a thermoplastic polymer and at least one phase of an elastomer, one of which is a continuous “matrix” phase in which the other phase is dispersed. Preferably, the

thermoplastic polymer functions as the continuous matrix phase with at least one discontinuous elastomer phase dispersed therein. As understood by those of ordinary skill in the art, which phase is the continuous phase depends upon the ratio of the thermoplastic to the elastomer phase as well as the viscosity ratios of the materials comprising the two phases. The ratio of thermoplastic to elastomer components in the TPE varies depending on the intended application. The selection of the types and amounts of these components is understood by those of ordinary skill in the art.

Preferably the thermoplastic and elastomer components of the TPE are selected such that they are generally compatible. "Generally compatible" means that the one phase (e.g., the elastomer phase of the TPE) has a good adhesion to and is finely dispersed in a continuous phase of the other component (e.g., the thermoplastic phase of the TPE). While not meant to limit the scope of the invention, the average elastomer particle size in generally compatible systems can range from as small as physically possible to about 100 μm in diameter. Desirably, the particle size of the elastomer particles can range from about 0.1 μm to about 5 μm in diameter, and preferably from about 0.05 μm to about 2 μm in diameter in particularly preferred compatible TPE systems.

Thermoplastics are generally materials that can be molded or otherwise shaped and reprocessed at temperatures at least as great as their softening or melting point. Polyolefins are preferred thermoplastic materials. As such, one particularly preferred TPE is a thermoplastic olefin elastomer (TPOE). TPOEs comprise at least one thermoplastic polyolefin and at least one elastomer.

Polyolefins are a fundamental building block in polymer science and engineering because of their low cost, high volume production based on petrochemical production. Non-limiting examples of polyolefins useful as thermoplastic olefins of the invention include homopolymers and copolymers of lower α -olefins such as 1-butene, 1-pentene, 1-hexene, 2-methyl-1-propene, 3-methyl-1-pentene, 4-methyl-1-pentene, and 5-methyl-1-hexene, as well as

ethylene, butylene, and propylene, with the homopolymer of propylene and copolymers of propylene being preferred. Most preferred is the homopolymer of propylene, polypropylene.

Beneficial effects of the invention are readily observed when
5 incorporating nucleating agents into TPEs comprising a propylene-based thermoplastic phase. Propylene-based materials, such as polypropylene, do not “self-nucleate.” That is, an external stimulus is required in order to provide nucleation sites for crystal growth propagation. Thus, there is a strong need to prevent the super cooling phenomenon witnessed more frequently with
10 materials that do not self-nucleate in order to promote rapid solidification of the crystallized structure therein to increase processing efficiency.

Polypropylene (PP) has thermoplastic properties best explained by a recitation of the following mechanical and physical properties: a rigid semicrystalline polymer with a modulus of about 1 GPa, a yield stress of about
15 35 MPa, and an elongation to ranging from about 10% to about 1,000 %. Selection of a polyolefin from commercial producers uses Melt Flow Index or Melt Flow Rate (MFI or MFR) properties. The MFI can range from about 0.05 to about 1400, and preferably from about 0.5 to about 70 g/10 min at 230°C under a 2.16 kg load. For polypropylene, that MFI should be from about 0.5 to
20 about 70 and preferably from about 1 to about 35 g/10 min at 230°C under a 2.16 kg load. Non-limiting examples of polypropylene useful for the present invention are those commercially available from multinational suppliers such as Dow Chemicals, Basell Polyolefins, and BP Amoco, Huntsman, etc.

Any suitable elastomer can form the elastomer phase of TPEs of the
25 invention. It is preferred that the elastomer has a substantially saturated hydrocarbon backbone chain that causes the copolymer to be relatively inert to ozone attack and oxidative degradation, but that the elastomer also has side-chain unsaturation available for curing. For example, natural rubber, polyisoprene rubber, styrenic copolymer elastomers (i.e., those elastomers

derived from styrene and at least one other monomer, elastomers that include styrene-butadiene (SB) rubber, styrene-ethylene-butadiene-styrene (SEBS) rubber, styrene-ethylene-propylene-styrene (SEPS) rubber, styrene-isoprene-styrene (SIS) rubber, styrene-ethylene-ethylene-propylene-styrene (SEEPS) rubber, styrene propylene-styrene (SPS) rubber, and others, all of which may optionally be hydrogenated), polybutadiene rubber, nitrile rubber, butyl rubber, ethylene-propylene-diene rubber (EPDM), ethylene-octene copolymers, and other elastomers are non-limiting examples of useful elastomers according to the invention.

10 Particularly preferred are styrenic copolymer elastomers and olefinic elastomers, especially EPDM, where the EPDM may be crosslinked partially or fully. Olefinic elastomers are especially useful in TPOEs because of their reasonable cost for properties desired. Of these elastomers, EPDM is preferred because it is a fundamental building block in polymer science and engineering
15 due to its low cost and high volume, as it is a commodity synthetic rubber since it is based on petrochemical production. EPDM encompasses copolymers of ethylene, propylene, and at least one nonconjugated diene. Most preferably, when EPDM comprises the elastomer phase of a TPE, the EPDM is crosslinked such that the TPE is a thermoplastic vulcanizate (TPV). The benefits of using
20 EPDM are best explained by the following mechanical and physical properties: low compression set, the ability to be oil extended to a broad range of hardness, and good thermal stability.

 Selection of an olefinic elastomer from commercial producers uses Mooney Viscosity properties. The Mooney Viscosity for olefinic elastomer can
25 range from about 1 to about 1,000, and preferably from about 20 to about 150 ML 1 + 4 @ 100°C. For EPDM, that Mooney Viscosity should be from about 1 to about 200, and preferably from about 20 to 70 ML 1 + 4 @ 100°C, when the elastomer is extended with oil. Non-limiting examples of EPDM useful for the present invention are those commercially available from multinational

companies such as Bayer Polymers, DuPont Dow Elastomers, Uniroyal Chemicals (now part of Crompton Corp.), ExxonMobil Chemicals, and others.

The elastomer itself may be provided in a variety of forms. For example, elastomers are available in liquid, powder, bale, shredded, or pelleted form. The form in which the elastomer is supplied influences the type of processing equipment and parameters needed to form the TPE. Those of ordinary skill in the art are readily familiar with processing elastomers in these various forms and will make the appropriate selections to arrive at the TPE component of the invention.

As noted above and unless proscribed otherwise by the appended claims, formation of TPE compositions comprising at least one nucleating agent are not necessarily limited to those based on any particular TPE and a wide variety of TPEs are commercially available. For example, PolyOne Corporation, Bayer, Crompton Corporation, DuPont Dow Elastomers, Teknor Apex, AES, Multibase, SoFTeR, Sumitomo, Asahi Kasei, Kraton, GLS, ExxonMobil Corporation, Uniroyal Chemical, and many other multinational companies have supplied commercial TPEs to the marketplace under an assortment of trade designations. These companies and many others provide a wide variety of TPEs that can be used in accordance with the present invention.

Nucleating Agent

A “nucleating agent” of the present invention is an additive that forms nuclei in a polymer melt, which nuclei initiate and promote crystal growth in at least one thermoplastic phase of the TPE. That is, the nucleating agent provides a nucleated TPE. It should be noted that the term “nucleating agent” is sometimes conventionally used in reference to a foaming process. It should be understood that those nucleating agents, to the extent that they are incapable of forming nuclei in a polymer melt for initiation and propagation of crystal growth are not encompassed within the definition of nucleating agents according to the present invention.

Advantageously, nucleating agents of the invention increase the number of sites where crystals are able to start to grow, which in turn reduces the overall size of crystals in the material. If the size of a crystal is reduced below the wavelength of light, the material becomes more transparent. In that
5 embodiment, the nucleating agent is sometimes referred to as a "clarifying agent." A further advantage of the use of nucleating agents to increase the number of nucleation sites in the thermoplastic phase of a TPE is a decrease in the time needed for a solid crystal structure to be obtained in the TPE, which allows the material to be processed at a higher temperature and facilitates faster
10 processing speeds.

The nucleating agent may be inorganic or organic. Suitable nucleating agents include, for example, nucleating talcs, titanium phosphate, glass powder, aluminum inorganics, benzoic acids and their aluminum or sodium salts (e.g., sodium benzoate), dibasic or monobasic acid salts, alicyclic acid salts,
15 polyacrylic acid derivatives (e.g., acrylic acid-grafted polypropylene), phosphate esters, and sorbitol derivatives (e.g., benzyldiene sorbitol).

Non-limiting examples of commercially available nucleating agents include those marketed under the trade designation, "Millad," by Milliken Chemicals (Spartanburg, SC). For example, Millad™ grades 3903, 3905, 3988,
20 3940, 3899, and 3908 are sorbitol derivatives sold by Milliken Chemicals and suitable for use as the nucleating agent in the present invention. Another suitable nucleating agent is Hyperform™ HPN-68 from Milliken Chemicals, which is specifically marketed as a new class of hyper nucleating agent that is stated to provide the fastest known crystallization rate in polypropylene while
25 enhancing the modulus of the article produced. Certain nucleating agents that function as clarifiers are produced by Roquette (France) and sold in the U.S. by Ciba Specialty Chemicals under the trade designation, IrgaClear™ (e.g., DM-LO). Similarly, Mitsui Toatsu Chemicals in Japan produces a clarifier that is sold in the U.S. by Mitsui Chemicals using the trade designation NC-4.

Other commercially available nucleating agents include polyacrylic acid-grafted polypropylene marketed under the trade designation, "Polybond," by Crompton Corporation – Uniroyal Chemical (Middlebury, CT). For example, Polybond™ 1001 is an acrylic acid-grafted polypropylene sold by Crompton Corporation that is suitable for use as the nucleating agent in the present invention. Still further commercially available nucleating agents include those organic phosphate ester salts marketed by Amfine Chemical Corporation (Allendale, NJ), such as those using the following trade designations – "ADKstab NA-11" and "ADKstab NA-21."

Other nucleating agents and clarifiers known to work in thermoplastic polymers are also suitable nucleating agents for this invention. For example, other suitable clarifiers and nucleating agents are summarized at the following website – www.plasticstechnology.com/articles/200207fa1.html.

Other Optional Additives

Additives, or further additives as compared to those in the base TPE, can optionally be included in the TPE compositions of the invention, if desired. The type and amount of additives so used will depend on the presence and amount of additives in the base TPE, which in large part depends on the source of that TPE. The type and amount of additives used will also depend on the desired application for which the end user will prepare articles from the TPE composition.

In any event, any suitable additive may be included in desired amounts in TPE compositions of the invention. For example, processing oils, compatibilizers, fillers (e.g., calcined clay, kaolin clay, nanoclay, talc, silicates, and carbonates), pigments and colorants (e.g., carbon black), flame retardants, antioxidants, conductive particles, UV-inhibitors, stabilizers, coupling agents (e.g., silanes and titanates), plasticizers, lubricants, antiblocking agents, antistatic agents, waxes, foaming agents, and combinations thereof may be

beneficially used in certain applications. Those of ordinary skill in the art will readily understand selection and use of such additives.

For instance, the use of processing oils to effect oil extension of the elastomer is well known in the art. In the production of TPEs, for example, it is often desirable to include an oil to extend the elastomer portion of the composition. This oil extension provides the property of lower hardness, while reducing cost of the elastomer to achieve the same TPE volume.

Oil can be a separate ingredient in the TPE composition or can be a part of the base elastomer itself, depending on the source of TPE. Non-limiting examples of oils suitable for optional use in the present invention include aromatic, paraffinic, and naphthathenic mineral oils. Both fully saturated oils as well as partially unsaturated oils are usable depending upon the particular end-use application and the type of elastomer selected for the formulation of the TPE. The concentration of oil in the TPE composition will generally range from about 0 to about 200 phr, and preferably from about 0 to about 100 phr, but may vary outside of these ranges as understood by those of ordinary skill in the art.

Further, the use of compatibilizers to promote integration of, for example, the discrete elastomer phase into the continuous thermoplastic phase (or vice versa) of a TPE is well known. In the production of TPEs, for example, it is often desirable and advantageous to include a compatibilizer to promote synergistic integration of the two distinct components – the thermoplastic and elastomer – when attempting to provide certain improved performance properties as compared to those observed with either component alone.

Preparation of Thermoplastic Elastomer (TPE) Compositions

Selection of Components

As used throughout the application, the phrase “thermoplastic elastomer composition” is used to refer to those compositions that include at least one

TPE and at least one nucleating agent. Optionally, the TPE composition may include other additives, such as those noted above.

The nucleating agent is used in an amount that promotes the rate of crystal growth propagation and crystal size desired. Preferably, the nucleating agent promotes production efficiency and provides other benefits as well. For example, when transparency of the TPE composition is important, minimizing the resulting crystal sizes is particularly beneficial.

While not meant to limit the scope of the invention, the amount of nucleating agent is preferably about 0.01% to about 5.0% by weight, more preferably about 0.1% to about 1.0% by weight, based on total weight of the base TPE. Of course, that amount can vary depending on the nature of the TPE and the proportion of thermoplastic and elastomer phases therein. When described in terms of the amount of thermoplastic present in the base TPE, the nucleating agent preferably comprises about 0.005% to about 5% by weight, more preferably about 0.1% to about 1.0% by weight, based on total weight of the thermoplastic phase in the TPE.

Selection of Processing Equipment and Processing of the TPE

Composition

As noted above, the introduction of the nucleating agent into the TPE compositions of the present invention is not complicated and can utilize any one of several methods: (1) addition during formation of the TPE itself, (2) addition as a post-processing step after the TPE itself is formed but before the TPE is transferred to another melt-processor (e.g., the end user who will transform the TPE composition into a desired article), or (3) addition during final processing of the TPE into a part or article. The third method is generally not preferred, not only because the end user typically desires to maximum their process efficiency, but also due to the possibility of the nucleating agent not being sufficiently dispersed within the TPE, resulting in non-uniform and unpredictable results.

The first method of adding the nucleating agent during formation of the TPE provides an in-situ formed TPE composition of the invention. According to this method, thermoplastic and elastomer components of at least one TPE are provided. At least one nucleating agent is then combined with the components
5 of the TPE. These components are then mixed, optionally in the presence of one or more optional additives. While not necessary, the components can also be heated in a further embodiment so that the thermoplastic phase of the TPE melts. In this further embodiment, the nucleating agent is able to substantially disperse within at least one thermoplastic phase of the TPE during formation of
10 the TPE itself. Formation of the TPE in this first method uses steps well familiar to those of ordinary skill in the art.

The second method for preparing a TPE composition incorporates at least one nucleating agent into at least one base TPE that has already been formed. According to this method, at least one base TPE is provided. At least
15 one nucleating agent is then combined with the TPE. These components are then mixed, optionally in the presence of one or more optional additives. While not necessary, the components can also be heated in a further embodiment so that the thermoplastic phase of the TPE melts. In this further embodiment, the nucleating agent is able to substantially disperse within at least one
20 thermoplastic phase of the TPE.

For either of these methods, processing of the TPE composition can occur via batch or continuous processing. Using either batch or continuous processing, components of the TPE composition can be mixed and heated to disperse the nucleating agent in the thermoplastic phase of the TPE in either a
25 single piece of equipment or in multiple pieces of equipment. Economies of scale for production lead to a preference for continuous processing. Further economies of scale can be obtained when further steps are continuously performed. For example, a particularly efficient process is one whereby the TPE compositions can be formed into desired shapes and sizes continuously
30 with their preparation.

In one embodiment of a batch process, TPE compositions can be prepared by mixing the components in a first piece of equipment. Mechanical mixers, such as Banbury-type, Brabender-type, roll mill, Buss, dry turbo mixers and the like are suitable for this purpose.

5 In one embodiment, all base components (i.e., TPE or components thereof, nucleating agent, and other optional additives, if used) of the TPE composition can be charged into the mixer. Mixing proceeds at any suitable pace to preferably effect a substantial mixing of the components. For example, mixing proceeds at a pace ranging from about 10 to about 100 rpm (revolutions
10 per minute), and preferably from about 75 to about 85 rpm for a duration ranging from about 1 to about 5 minutes, and preferably from about 2 to about 4 minutes in certain embodiments of the invention.

 If dispersion of the nucleating agent into the thermoplastic phase of the TPE is desired, heat can also be applied during this mixing step. When heat is
15 applied, it is applied at a temperature sufficient to achieve the desired effects. Preferably, the components are heated to a temperature sufficient to melt or soften the thermoplastic component of the TPE. For example, when the thermoplastic phase of the TPE comprises a propylene-based polymer, the components are typically heated to a temperature ranging from about 170°C to
20 about 210°C, and preferably from about 185°C to about 195°C. It is not necessary, however, for the nucleating agent to melt during this melt mixing step.

 In a batch process, the TPE composition is then transferred to other equipment for formation into the desired shape and size. Typically, this will be
25 a shape and size that enables an end user to melt process the TPE composition into the desired article. For example, plugs of the TPE composition can be removed from the mixer and compression molded into, for example, a 7.6 cm x 15.2 cm x 0.31 cm (3 in x 6 in x 0.125 in) plaque mold at a temperature ranging from about 170°C to about 230°C, and preferably from about 195°C to about

215°C. The plug material can be held under, for example, no pressure for 30 seconds, after which pressure can be increased to 1,100 kN force over a period of about 3 minutes. After application of pressure of 1,100 kN force for 4 minutes, the samples can be cooled to ambient temperature while pressure is maintained.

During continuous processing, the components can be first mixed in a suitable mixer. Mechanical mixers, such as Banbury-type, Brabender-type, roll mill, Buss, dry turbo mixers and the like are suitable for this purpose. Sufficient heat is generally generated in these mixers to melt mix the nucleating agent into the TPE composition. In this embodiment, the mixed components are then conveyed continuously to another piece of equipment, where the mixture is heated to form the TPE composition. Processing then continues by forming the TPE composition into the desired shape and size without the need to transfer the bulk contents to another piece of equipment during the continuous process. The TPE can even be formed into the end-use article, if desired, using continuous processing.

An example of a continuous process for forming the TPE composition comprises utilization of reactive extrusion equipment. A wide variety of reactive extrusion equipment can be employed in this manner. Preferred is a twin screw co-rotating extruder with a length-to-diameter (L/D) ratio ranging from about 38 to about 60, and preferably from about 40 to about 52.

Reactive extrusion allows for dynamic vulcanization of the elastomer phase of the TPE to occur, which is preferably when preparing thermoplastic vulcanizates (TPVs). Dynamic vulcanization, when used in conjunction with the present invention, can further advantageously reduce processing time and throughput. However, methods other than dynamic vulcanization can be utilized to prepare TPE compositions of the invention. For example, the elastomer component of the TPE can be cured in the absence of the thermoplastic component, powdered, and mixed with the thermoplastic

component at a temperature above the melting or softening point of the thermoplastic component to form a TPE.

While optional, a TPE may be fully or partially crosslinked. As noted above, in certain embodiments of the invention, it is preferred that the TPE composition be a TPV. To achieve reaction, and hence crosslinking (also referred to as "curing") of a TPE so that it becomes a TPV, the mixture is typically heated to a temperature substantially equal to or greater than the softening or melting point of any thermoplastic component of the TPE and for a sufficient time to obtain a composition of the desired homogeneity and crosslinking of the elastomer phase. For example, the extrusion profile for a preferred polypropylene (PP)/EPDM reactive extrusion can be a flat 190°C profile and 500 rpm. The reaction components can be fed into the reaction extruder at 27 kg/hr (60 lbs/hr) using, for example, a 25-mm twin screw extruder.

The components of the TPE composition may be added to the processing equipment in any suitable amount and in any suitable order. As noted above, the nucleating agent can be added before or after formation of the base TPE. Those of ordinary skill in the art will readily recognize the ability to vary the amount and order of addition of the components within the TPE composition in general.

Usefulness of the Invention

TPE compositions of the invention can be formed into a variety of articles as well understood by those of ordinary skill in the art. For example, TPE compositions can be reprocessed, such as by being pressed, compression molded, injection molded, calendared, thermoformed, blow-molded, or extruded into final articles. When reprocessing TPE compositions of the invention, the composition is generally heated to a temperature of at least the softening or melting point of the thermoplastic component of the TPE composition in order

to facilitate further forming into desired articles of various shapes and sizes. The end user of the TPE compositions will benefit by the processing advantages described throughout. Further, the present invention promotes the end user's ability to enhance transparency of certain articles formed from the TPE

5 compositions.

Further embodiments of the invention are described in the following non-limiting example.

Examples

10 General Experimental Materials Examples

Unless noted otherwise, the following materials were used as indicated in the examples described below.

Table 1		
Trade Designation or Abbreviation	Description	Source
ADKstab NA-11	Clarifier	Amfine Chemical Corporation; Allendale, NJ (www.amfine.com)
Forprene™ 6EK5 01 A65	High-performance thermoplastic elastomer, natural, UV-stabilized	PolyOne Corporation; Avon Lake, OH (www.polyone.com)
Forprene™ 6M0 901 A70	Thermoplastic elastomer	PolyOne Corporation; Avon Lake, OH (www.polyone.com)
Forprene™ 6M0 901 A75	Thermoplastic elastomer	PolyOne Corporation; Avon Lake, OH (www.polyone.com)
Forprene™ 6M0 901 A85	Thermoplastic elastomer	PolyOne Corporation; Avon Lake, OH (www.polyone.com)
Millad™ 3905	Sorbitol derivative clarifier	Milliken Chemical; Spartanburg, SC
Hyperform™ HPN-68	Norbornane carboxylic acid salt	Milliken Chemical; Spartanburg, SC
Millad™ 3908	Sorbitol derivative clarifier	Milliken Chemical; Spartanburg, SC
Polybond™ 1001	Acrylic acid-grafted polypropylene	Crompton Corporation – Uniroyal Chemical; Middlebury, CT (www.cromptoncorp.com)
Santoprene™ 101-73	Thermoplastic elastomer	Advanced Elastomer Systems, L.P.; Akron, OH (www.santoprene.com)
Santoprene™ 101-87	Thermoplastic elastomer	Advanced Elastomer Systems, L.P.; Akron, OH (www.santoprene.com)
Sodium Benzoate	Sodium Benzoate	Velsicol Chemical Corp. (Rosemont, IL) or other suppliers

Comparative Example C1 and Examples 1-3

In a Brabender-type mixer, the Forprene™ 6EK5 01 A65 listed in Table 2 was mixed at a temperature of 190°C and a speed of 60 r.p.m. When the

5 Forprene™ 6EK5 01 A65 began to flux as understood by those of skill in the art, the remaining materials listed in Table 2 were then added to the mixer. Mixing continued for 30 seconds. Then, the material was pressed into plaques using a compression molder operating at a temperature of 205°C (400°F).

Table 2				
Material	Parts by Weight (%)			
	Comparative Example C1	Ex. 1	Ex. 2	Ex. 3
Forprene™ 6EK5 01 A65	100.00	99.90	99.90	99.90
Sodium Benzoate	-	0.10	-	-
Millad™ 3908	-	-	0.10	-
ADKstab NA-11	-	-	-	0.10

The composition without a nucleating agent (Comparative Example C1) had a second melt crystallization temperature of 109°C, which was considerably lower than those of compositions of the invention. Compositions of the invention (Example 1-3) displayed second melt crystallization temperatures of 121°C, 123°C, and 124°C, respectively. Thus, processing efficiency is improved when using compositions of the present invention, as a solid crystal structure is imparted more rapidly therein upon cooling.

Comparative Example C2 and Examples 4-9

In a Brabender-type mixer, the Forprene™ 6M0 901 A75 listed in Table 3 was mixed at a temperature of 190°C and a speed of 60 r.p.m. When the Forprene™ 6M0 901 A75 began to flux as understood by those of skill in the art, the remaining materials listed in Table 3 were then added to the mixer. Mixing continued for 30 seconds. Then, the material was pressed into plaques using a compression molder operating at a temperature of 205°C (400°F).

Table 3							
Material	Parts by Weight (%)						
	Comparative Example C2	Ex. 4	Ex. 5	Ex. 6	Ex. 7	Ex. 8	Ex. 9
Forprene™ 6M0 901 A75	100.00	99.75	99.75	99.75	99.75	99.90	99.65
Sodium Benzoate	-	0.25	-	-	-	-	0.10
Millad™ 3908	-	-	0.25	-	-	-	0.10
ADKstab NA-11	-	-	-	0.25	-	-	-
Polybond™ 1001	-	-	-	-	0.25	0.10	0.15

As with the previous examples, the second melt crystallization temperatures for the compositions containing a nucleating agent (Examples 4-9) were higher than that composition without a nucleating agent (Comparative Example C2), which is consistent with the compositions of the present invention providing for improved processing efficiency.

Examples 10-11 and Comparative Examples C3-C5

Two commercially available TPVs were modified according to the present invention for use in two different applications. A part was formed for each application, the type and thickness of which is reported in Table 4 below. Also noted in Table 4 is the commercially available TPV that was so modified according to each of Examples 10-11.

The modified TPV (i.e., that containing a nucleating agent) formed was relatively fast molding (denominated herein as a "Fast Molding Grade") as compared to unmodified commercially available TPVs having a similar hardness value. Those unmodified commercially available TPVs and their cycle times are noted in Table 5 below.

The fast molding grades of Example 10 and 11 were prepared using a combination of 0.1% sodium benzoate, 0.1% ADKstab NA-11, 0.2% nucleating talc, and 0.15% Hyperform™ HPN-68 as a nucleating agent, all weights of which were based on total weight of the base TPV.

Table 4			
Example No.	Application/Relevant Dimension	TPV Modified to Produce Fast Molding Grade	Cycle Time of Fast Molding Grade (sec)
10	Strain Relief Part (Wall Thickness = 0.3175 cm (0.125 in))	Forprene™ 6M0 901 A75	24
11	Grip (Wall Thickness = 0.9525 cm (0.375 in))	Forprene™ 6M0 901 A85	71

Table 5				
Comparative Example No.	Application/Relevant Dimension	Unmodified TPV Grade	Cycle Time of Unmodified TPV Grade (sec)	% Reduction in Cycle Time
C3	Strain Relief Part (Wall Thickness = 0.3175 cm (0.125 in))	Santoprene™ 101-73	31	22.6 (Ex. 10)
C4	Strain Relief Part (Wall Thickness = 0.3175 cm (0.125 in))	Forprene™ 6M0 901 A70	31	22.6 (Ex. 10)
C5	Grip (Wall Thickness = 0.9525 cm (0.375 in))	Santoprene™ 101-87	91	22.0 (Ex. 11)

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As can be seen when comparing Tables 4 and 5, in all cases, the new Fast Molding Grade of the present invention enabled lower cycle times when used in an injection molding production line as compared to the cycle time when a comparable, unmodified commercially available TPV of a similar hardness value was similarly used (i.e., comparing the Fast Molding Grade of Example 10 to Comparative Examples C3 and C4 and comparing the Fast Molding Grade of Example 11 to Comparative Example C5). Thus, advantageously, the use of the fast molding grades resulted in higher productivity as compared to a comparable commercially available TPV with a

similar hardness value. The overall percentage decrease in cycle time is reported in Table 5 for each comparison.

- 5 Various modifications and alterations of the invention will become apparent to those skilled in the art without departing from the spirit and scope of the invention, which is defined by the accompanying claims. It should be noted that steps recited in any method claims below do not necessarily need to be performed in the order that they are recited. Those of ordinary skill in the art will recognize variations in performing the steps from the order in which they are recited.